organic papers

Acta Crystallographica Section E Structure Reports Online

ISSN 1600-5368

Chun-Hua Diao* and Zhi Fan

College of Sciences, Tianjin University of Science and Technology, Tianjin 300222, People's Republic of China

Correspondence e-mail: diao_chunhua@163.com

Key indicators

Single-crystal X-ray study T = 294 K Mean σ (C–C) = 0.003 Å R factor = 0.044 wR factor = 0.131 Data-to-parameter ratio = 15.4

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

4-[(*E*)-4-(2-{4-[(*E*)-(1,5-Dimethyl-3-oxo-2-phenyl-2,3-dihydro-1*H*-pyrazol-4-yl)iminomethyl]phenoxy}ethoxy)benzylideneamino]-1,5-dimethyl-1-phenyl-1*H*-dihydropyrazol-3(2*H*)-one

In the title compound, $C_{38}H_{36}N_6O_4$, a crystallographic center of symmetry is located at the mid-point of the central C–C bond. The 4-hydroxybenzaldehyde residue makes dihedral angles of 8.09 (9) and 68.93 (7)°, respectively, with the pyrazolone ring and the terminal phenyl ring. The crystal packing is stabilized by intermolecular C–H···O hydrogen bonds that link molecules into one-dimensional extended chains.

Comment

There has been steady growth of interest in the structure and reactivity of Schiff bases owing to their potential biological activities, such as antibacterial and antitumor (Klayman et al., 1979). Consequently, many Schiff base derivatives have been synthesized and employed to develop protein and enzyme mimics, such as models to mimic hydrolase in the hydrolysis of p-nitrophenyl picolinate (Li et al., 2005). Among the large number of such compounds, 4-amino-1,5-dimethyl-2-phenylpyrazol-3-one forms a variety of Schiff bases with aldehydes, and the synthesis and crystal structures of some of them, such as (E)-1,5-dimethyl-4-{2-[2-(2-nitrophenoxy)-ethoxy]benzylideneamino}-2-phenyl-1H-pyrazol-3(2H)-one (Diao & Chen, 2006) and (E)-4-[3-ethoxy-4-(2-phenoxyethoxy)benzylideneamino]-1,5-dimethyl-2-phenyl-1H-pyrazol-3(2H)-one (Zhang et al., 2006), have been reported. We report here the synthesis and structure of the title compound, (I).



In (I) (Fig. 1), a crystallographic center of symmetry is located at the mid-point of the central $C1-C1^i$ bond [symmetry code: (i) -x + 2, -y, -z]. The bond lengths and angles are within normal ranges (Allen *et al.*, 1987). The pyrazolone ring (C9-C11/N1-N3/O2) is almost planar, with an r.m.s. deviation for fitted atoms of 0.0475 Å. It makes a dihedral angle of 60.87 (7)° with its attached phenyl ring (C14-C19). The 4-hydroxybenzaldehyde residue (C2-C8/O1) is essentially planar, with an r.m.s. deviation for the fitted atoms of 0.0127 Å. This plane makes dihedral angles of 8.09 (9) and 68.93 (7)°, respectively, with the pyrazolone ring (C9-C11/N1-N3/O2) and the terminal (C14-C19) phenyl ring.

© 2006 International Union of Crystallography All rights reserved Received 15 September 2006 Accepted 18 September 2006

Z = 2

 $D_x = 1.288 \text{ Mg m}^{-3}$

 $0.38 \times 0.26 \times 0.20$ mm

9053 measured reflections

3383 independent reflections 1892 reflections with $I > 2\sigma(I)$

Mo $K\alpha$ radiation

 $\mu = 0.09 \text{ mm}^{-1}$

T = 294 (2) K

Block, yellow

 $R_{\rm int} = 0.040$

 $\theta_{\rm max} = 26.5^{\circ}$



Figure 1

The molecular structure of (I), with displacement ellipsoids for non-H atoms drawn at the 30% probability level [symmetry code: (I) -x + 2, -y, -z].





The crystal packing is stabilized by weak non-classical intermolecular $C-H\cdots O$ hydrogen bonds (Table 1). These $C-H\cdots O$ hydrogen bonds link molecules into one-dimensional extended chains (Fig. 2).

Experimental

An anhydrous ethanol solution (100 ml) of 4-[2-(4-formyl-phenoxy)ethoxy]benzaldehyde (2.70 g, 10 mmol) was added to an anhydrous ethanol solution (50 ml) of 4-amino-1,5-dimethyl-2phenylpyrazol-3-one (2.03 g, 10 mmol) and the mixture was stirred at 350 K for 5 h under nitrogen, giving a yellow precipitate. The product was isolated, recrystallized from acetonitrile and then dried in a vacuum to give the pure compound in 82% yield. Yellow single crystals of (I) suitable for X-ray analysis were obtained by slow evaporation of an N,N-dimethylformamide solution.

Crystal data

 $\begin{array}{l} C_{38}H_{36}N_6O_4 \\ M_r = 640.73 \\ \text{Monoclinic, } P2_1/c \\ a = 13.199 \ (5) \ \text{\AA} \\ b = 6.802 \ (3) \ \text{\AA} \\ c = 18.841 \ (7) \ \text{\AA} \\ \beta = 102.315 \ (7)^\circ \\ V = 1652.6 \ (11) \ \text{\AA}^3 \end{array}$

Data collection

Bruker SMART APEX CCD areadetector diffractometer φ and ω scans Absorption correction: multi-scan (*SADABS*; Sheldrick, 1996) $T_{min} = 0.951, T_{max} = 0.983$

Refinement

D-C7

Refinement on F^2 w $R[F^2 > 2\sigma(F^2)] = 0.044$ w $wR(F^2) = 0.131$ SS = 1.00(a3383 reflections Δ 219 parameters Δ H-atom parameters constrained

$w = 1/[\sigma^2(F_0^2) + (0.0603P)^2]$
+ 0.1695P]
where $P = (F_0^2 + 2F_c^2)/3$
$(\Delta/\sigma)_{\rm max} < 0.001$
$\Delta \rho_{\rm max} = 0.17 \text{ e } \text{\AA}^{-3}$
$\Delta \rho_{\rm min} = -0.18 \ {\rm e} \ {\rm \AA}^{-3}$

l'able l			
Hydrogen-bond	geometry	(Å,	°).

-H···A	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - \mathbf{H} \cdot \cdot \cdot A$
$-H7\cdots O2^{ii}$	0.93	2.50	3.332 (3)	150

Symmetry code: (ii) -x + 1, -y, -z.

H atoms were included in calculated positions and refined using the riding model approximation: C-H = 0.93 Å and $U_{iso}(H) = 1.2U_{eq}(C)$ for aromatic C-H; C-H = 0.97 Å and $U_{iso}(H) = 1.2U_{eq}(C)$ for methylene CH₂; C-H = 0.96 Å and $U_{iso}(H) = 1.5U_{eq}(C)$ for methyl CH₃.

Data collection: *SMART* (Bruker, 1999); cell refinement: *SAINT* (Bruker, 1999); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997*a*); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997*a*); molecular graphics: *SHELXTL* (Sheldrick, 1997*b*); software used to prepare material for publication: *SHELXTL*.

References

- Allen, F. H., Kennard, O., Watson, D. G., Brammer, L., Orpen, A. G. & Taylor, R. (1987). J. Chem. Soc. Perkin Trans. 2, pp. S1–19.
- Bruker (1999). SMART (Version 5.0) and SAINT (Version 4.0) for Windows NT. Bruker AXS Inc., Madison, Wisconsin, USA.
- Diao, C.-H. & Chen, X. (2006). Acta Cryst. E62, 04422-04424.
- Klayman, D. L., Bartosevich, J. F., Griffin, T. S., Mason, C. J. & Scovill, J. P. (1979). J. Med. Chem. 22, 855–862.
- Li, J.-Z., Xu, B., Li, S.-X., Zeng, W. & Qin, S.-Y. (2005). *Transition Met. Chem.* **30**, 669–676.
- Sheldrick, G. M. (1996). SADABS. University of Göttingen, Germany.
- Sheldrick, G. M. (1997a). SHELXS97 and SHELXL97. University of Göttingen, Germany.
- Sheldrick, G. M. (1997b). SHELXTL. Version 5.10 for Windows NT. Bruker AXS Inc., Madison, Wisconsin, USA.
- Zhang, W.-J., Duan, Z.-Y. & Zhao, X. (2006). Acta Cryst. E62, o2963o2964.